

## Chapter 10 Measurement Of Matter

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Ex. 1 dozen = 12 Chapter 10 10.1 The Mole: A Measurement of Matter Chemist use a unit that is a specified unit of particles. This unit is called the mole.

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10.1 The Mole: A Measure-ment of Matter 26 > Copyright © Pearson Education, Inc., or its affiliates. All Rights Reserved.. What are the different ways you can

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At STP, one mole, or  $6.02 \times 10^{23}$  particles of any gas occupies a volume of 22.4 L or  $22.4 \text{ dm}^3$ , used to convert between the number of moles of gas and the volume of gas at STP  $1 \text{ L} = 1 \text{ dm}^3$

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If a 10.00 g sample is placed in a graduated cylinder that contains 15.0 mL of water, what is the total volume that would be occupied? At  $20^\circ\text{C}$ , the density of fresh water is  $0.9982 \text{ kg/m}^3$ , and the density of seawater is  $1.025 \text{ kg/m}^3$ .

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