

## Chapter 20 Worksheet Redox

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Chapter 20 Worksheet Redox Chapter 20 Worksheet: Redox I Determine what is oxidized and what is reduced in each reaction Identify the oxidizing agent and the reducing agent, also 1  $2\text{Sr} + \text{O}_2$  2  $2\text{Li} + \text{S}$  3  $2\text{Cs} + \text{Br}_2$  4  $3\text{Mg} + \text{N}_2$  5  $4\text{Fe} + 3\text{O}_2$  6  $2\text{Fe}_2\text{O}_3$  6  $\text{Cl}_2 +$

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Correct WS 17.1 Exercises from Chapter 20 Finish Electrochemistry Lab #20 Data Only HW: "17.2 Exercises" Odd's only from chapter 20. Electrochemistry Lab #20 NOT homework, but Intro, objectives, prelab, & data analysis will be due next week.

**Milstead, Millie / AP Chemistry Ch. 20 Electrochemistry**

Chapter 20 Worksheet: Redox I. Determine what is oxidized and what is reduced in each reaction. Identify the oxidizing agent and the reducing agent, also. 1.  $2\text{Sr} + \text{O}_2 \rightarrow 2\text{SrO}$  2.  $2\text{Li} + \text{S} \rightarrow \text{Li}_2\text{S}$  3.  $2\text{Cs} + \text{Br}_2 \rightarrow 2\text{CsBr}$  4.  $3\text{Mg} + \text{N}_2 \rightarrow \text{Mg}_3\text{N}_2$  5.  $4\text{Fe} + 3\text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3$  6.  $\text{Cl}_2 + 2\text{NaBr} \rightarrow 2\text{NaCl} + \text{Br}_2$  7.  $\text{Si} + 2\text{F}_2 \rightarrow \text{SiF}_4$  8.

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